

Chromatographic Protein Purification

Protein Property	Chromatography
Size	Gel filtration (GFC) Gel permeation (GPC) Size exclusion (SEC)
Charge	Ion exchange (IEC)
Hydrophobicity	Hydrophobic interaction (HIC) [Reverse phase (RPC)]
Specific binding	Affinity (AC)

Size-Exclusion Chromatography

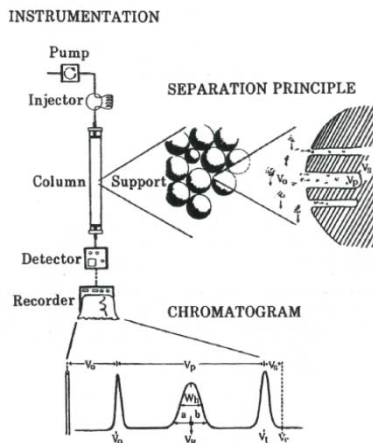
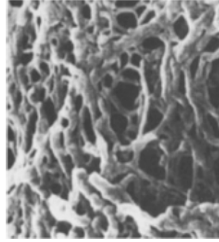
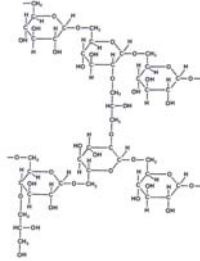


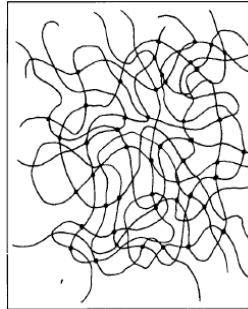
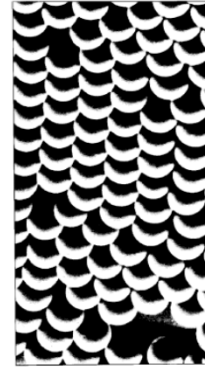
Figure 3-1. Fundamentals of gel filtration. Solutes injected into the column are separated according to decreasing size due to incompatibility between the solute dimension and the pore size of the support. V_0 = void volume between the support particles, V_p = pore volume and V_s = matrix volume of the support. V_r = elution volume of the solute, V_t = total liquid volume of the column and V_c = total geometric volume of the column. Column plate number $N = 5.54 \times (V_r/w_a)^2$ where w_a is the peak width at half peak height and b/a = peak symmetry at 10% peak height.

Size-Exclusion Chromatography

Sephadex

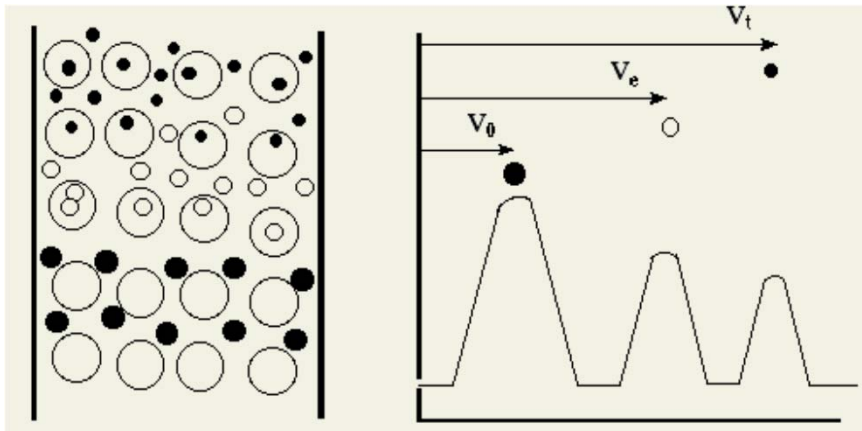


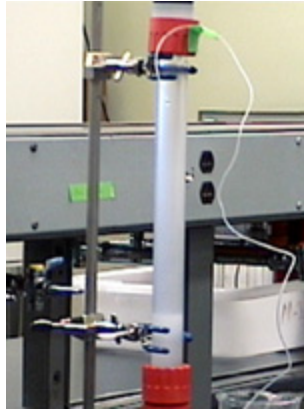
Scanning electro micrograph of 2% agarose gel



- Sephadex is a gel consisting of crosslinked dextran molecules. Dextran is a branched polymer of glucose molecules.
- Sephadex is formed into beads (20-300 μm), and each bead thus comprises a 3D network of pores into which solute molecules, such as proteins, can diffuse.

Size-Exclusion Chromatography





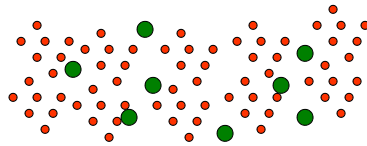
Fancy



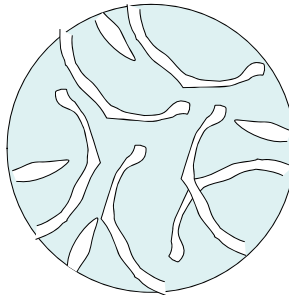
Plain
4°C (cold room)

70

**Molecular sieve chromatography
(= gel filtration, Sephadex chromatography)**



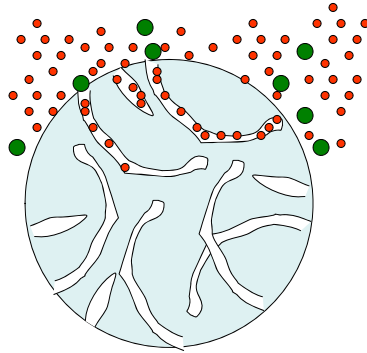
Sephadex bead



71

Molecular sieve chromatography

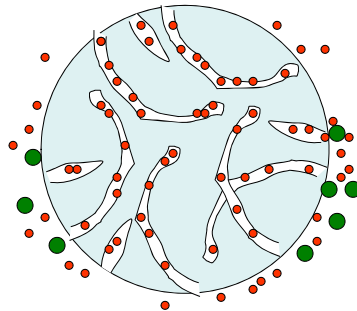
Sephadex bead



72

Molecular sieve chromatography

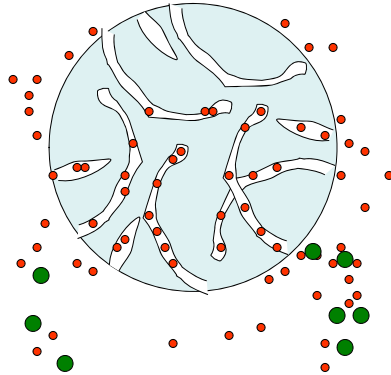
Sephadex bead



73

Molecular sieve chromatography

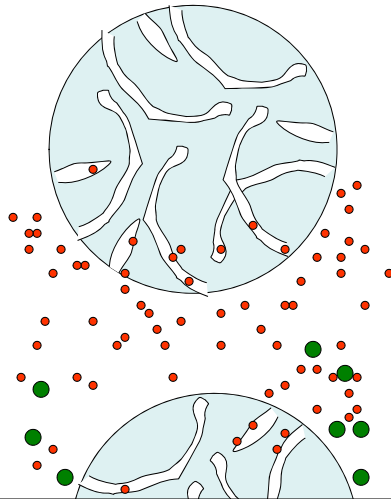
Sephadex bead



74

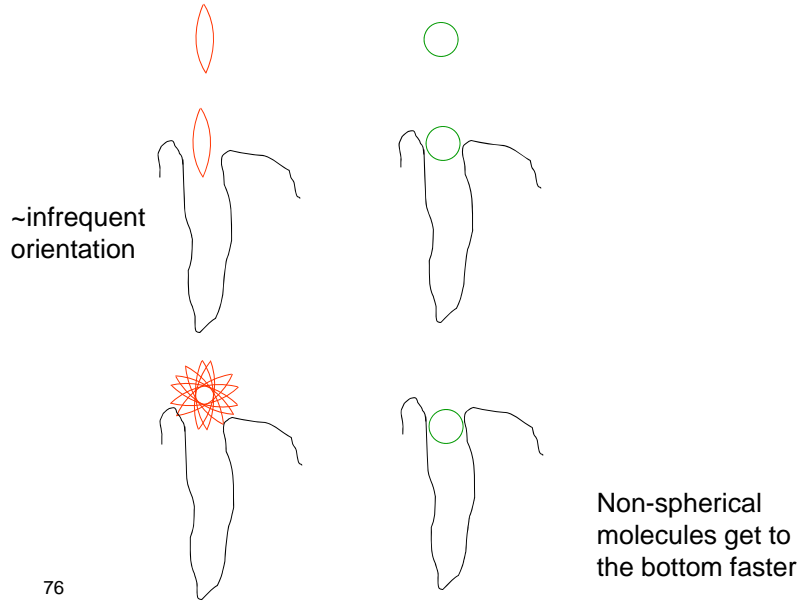
Molecular sieve chromatography

Sephadex bead



75

Larger molecules get to the bottom faster, and
 Non-spherical molecules get to the bottom faster



Analysis of Chromatogram

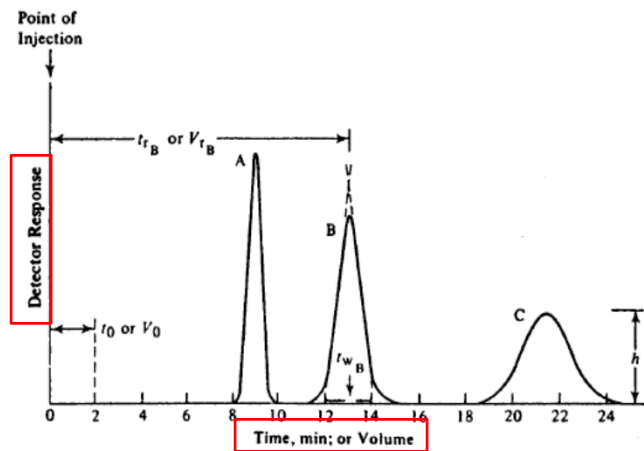
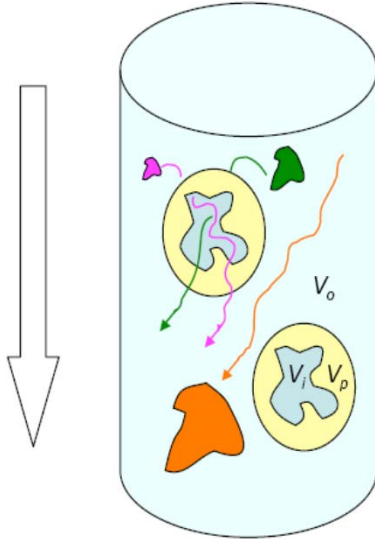


FIGURE 21.2. Chromatogram of the three-component mixture of Figure 21.1. t_0 = time for solvent to traverse the column, t_{r_B} = retention time of substance B, t_{w_B} = peak basewidth of substance B, h = peak height. Units can also be given in terms of volume rather than time: V_0 , V_{r_B} , V_{w_B} , and so forth.

Analysis of Chromatogram



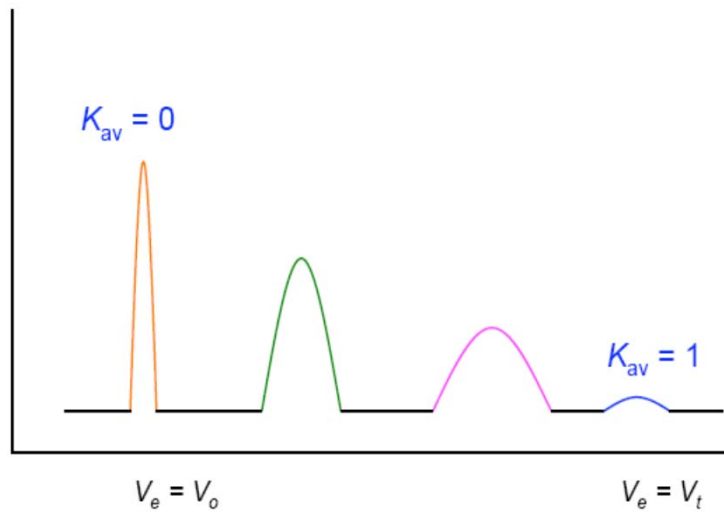
$$V_{\text{total}} = V_0 (= 0.35 \times V_t) + V_p + V_i$$

$$K_{\text{av}} = \frac{V_e - V_0}{V_i + V_p} = \frac{V_e - V_0}{V_t - V_0}$$

$$0 \leq K_{\text{av}} \leq 1$$

$$K_{\text{av}} = -a \times \log \text{MW} + b$$

Analysis of Chromatogram



Kav vs. log MW

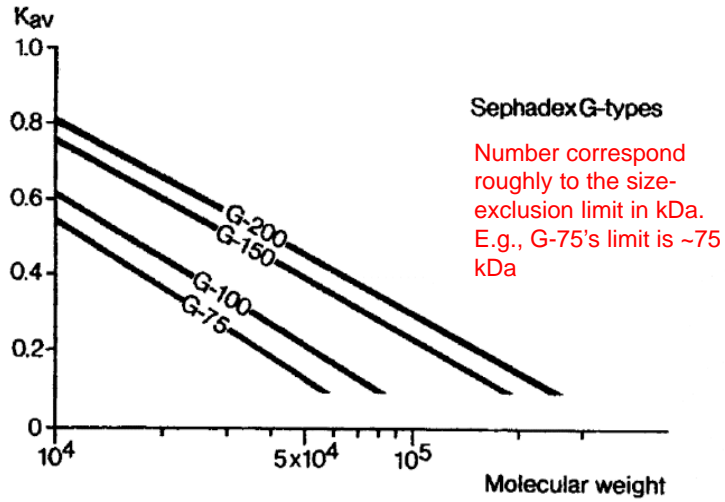
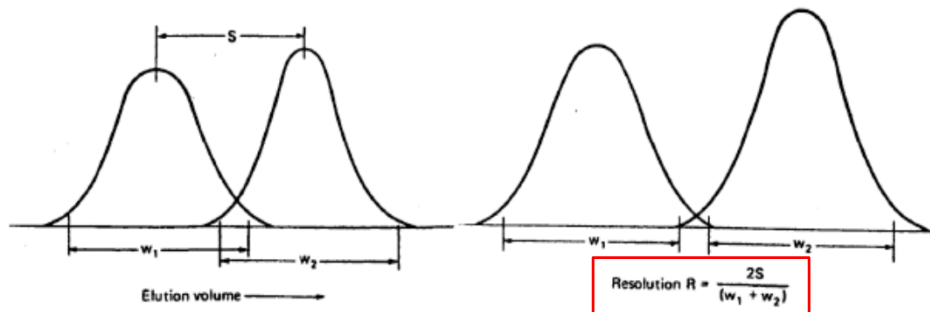


Fig. 24. Selectivity curves of Sephadex G-types, globular proteins.

Chromatographic Resolution



Chromatographic Resolution

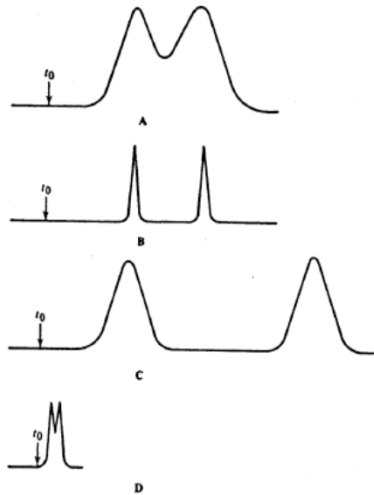
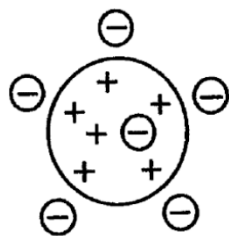
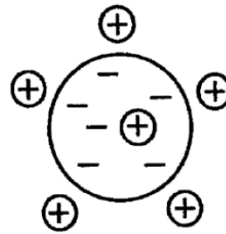


FIGURE 21.7. Effect of selectivity, efficiency, and capacity factor on resolution. A: Poor resolution. B: Good resolution due to column efficiency. C: Good resolution due to column selectivity. D: Poor resolution due to low capacity factor despite adequate column efficiency and selectivity. Courtesy of Varian Associates.

Ion Exchange Resins w/ exchangeable counter ions



ANION exchanger with
exchangeable counter-ions



CATION exchanger with
exchangeable counter-ions

Functional groups used in Ion-exchange resin

TABLE 4-2
Ion Exchange Groups Used in the Purification of Proteins

FORMULA	NAME	ABBREVIATION
Strong anion		
$-\text{CH}_2\text{N}^+(\text{CH}_3)_3$	trimethylaminoethyl	TAM
$-\text{C}_2\text{H}_4\text{N}^+(\text{C}_2\text{H}_5)_3$	triethylaminoethyl	TEAE
$-\text{C}_2\text{H}_4\text{N}^+(\text{C}_2\text{H}_5)_2\text{CH}_2-\text{CH}(\text{OH})\text{CH}_3$	diethyl-2-hydroxypropylamino-ethyl	QAE
Weak anion		
$-\text{C}_2\text{H}_4\text{N}^+\text{H}_3$	aminoethyl	AE
$-\text{C}_2\text{H}_4\text{N}^+\text{H}(\text{C}_2\text{H}_5)_2$	diethylaminoethyl	DEAE
Strong cation		
$-\text{SO}_3$	sulpho	S
$-\text{CH}_2\text{SO}_3$	sulphomethyl	SM
$-\text{C}_3\text{H}_6\text{SO}_3$	sulphopropyl	SP
Weak cation		
$-\text{COO}^-$	carboxy	C
$-\text{CH}_2\text{COO}^-$	carboxymethyl	CM

