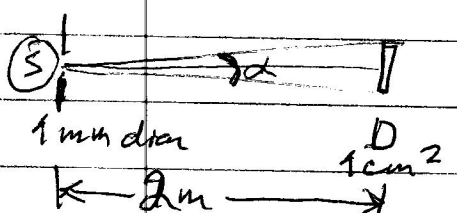


Chem 524 - Homework #1 Answers 2013
plus some old ones

2-4 a. Flux is 3.18×10^{15} ph/s $\lambda = 632.8$ nm
(cont of like 2-12 - attached) $E_p = h\nu = hc/\lambda$
 $= 3.14 \times 10^{-19}$ J
 $\Phi = 3.18 \times 10^{15} \times 3.14 \times 10^{-19} = 9.98 \times 10^{-4}$ W ~ 1 mW
b. $E = \Phi/A = 9.98 \times 10^{-4}$ W / $5 \text{ mm}^2 \cdot (10^{-2} \text{ cm}^2/\text{mm}^2) = 2 \times 10^{-7}$ W/cm²
c. $E^0 = 3.18 \times 10^{15} / 0.05 \text{ cm}^2 = 6.36 \times 10^{16}$ ph/s/cm²

2-7 - basically same idea as 2-3, 2-14 attached



$R = A/d^2 = 1 \text{ cm}^2 / (200 \text{ cm})^2 = 2.5 \times 10^{-5}$
 $B = 2 \text{ W cm}^{-2} \text{ s}^{-1} \text{ nm}^{-1} \cdot 2.5 \times 10^{-5} \text{ s} \cdot (0.05)^2 \text{ cm}^2$
 $= 3.93 \times 10^{-7} \text{ W nm}^{-1}$
 $E = 10 \text{ nm} (6.6 \times 10^{-19} \text{ J/ph})^{-1} (3.93 \times 10^{-7} \text{ W nm}^{-1}) E_p = \frac{hc}{\lambda} = 6.6 \times 10^{-19}$
 $= 5.9 \times 10^{12}$ phat/sec

2-15 if transmit 90% \Rightarrow absorb 10% = 0.1 μ W
= luminescence would be zero Stokes shift
(i.e. lum at same λ as exc - not happen)
add 100% quantum yield (no loss)

new
2-16 population - do relative pop $v=0$ and $v=1$
 $P_{01}^{1000} = (n_1/n_0) = e^{-\Delta E/kT} = \exp\left(\frac{-3600}{1000 \cdot 2/3}\right) \Delta E_{10^2} = 3600 \text{ cm}^{-1}$
 $= 0.0045$ $= 16.5 \text{ J cm}^{-1}$
 $P_{01}^{300} = \exp\left(\frac{-3600}{300 \cdot 2/3}\right) = e^{-18} = 1.5 \times 10^{-8}$ $kT \sim 2/3 \text{ cm}^{-1}/\text{K}$
 $P_{01}^{1000} = \exp\left(\frac{-1650}{1000 \cdot 2/3}\right) = 8.4 \times 10^{-2}$
 $P_{01}^{300} = \exp\left(\frac{-1650}{300}\right) = e^{-5.5} = 2.16 \times 10^{-4}$

extra!

Chem 524—~~2009~~²⁰¹¹ Homework Set #1, Some answers

Part I INTRODUCTION

Chapter 2

2-3

#3 An extended source is spherical in shape with a radius of 2.00 cm and emits 12.57 W from 399.5 to 400.5 nm. Determine the spectral radiant emittance, the spectral radiant intensity, and the spectral radiance at 400nm.

Answer:

$$\text{Radiant emittance: } M = \Phi / A = 12.57\text{W} / [4 \times \pi \times (2.00\text{cm})^2] = 12.57\text{W} / (4 \times 12.57\text{cm}^2) = 0.25 \text{ Wcm}^{-2}$$

$$\text{Radiant intensity: } I = \Phi / \Omega = 12.57\text{W} / 12.57\text{sr} = 1.00 \text{ Wsr}^{-1}$$

$$\text{Radiance: } B = \Phi / (\Omega A) = 12.57\text{W} / [12.57\text{sr} \times 4 \times \pi \times (2.00\text{cm})^2] = 2.00 \times 10^{-2} \text{ Wsr}^{-1}\text{cm}^{-2}$$

15 A sample is illuminated with 1.0 μ W of radiation and transmits 90% of the incident radiation. Calculate the maximum value of the luminescence radiant power that could be observed.

Answer:

$$\Phi_L = k(\Phi_0 - \Phi)$$

$$\text{When } k=1, \Phi_{L,\text{max}} = \Phi_0 - \Phi = 1.0 \mu\text{W} - 90\% \times 1.0 \mu\text{W} = 0.1 \mu\text{W}$$

2-16

16 The irradiance on a receptor 50cm from a source with a projected area of 0.010cm² is 2.0 μ W cm⁻². Calculate the radiance of the source.

& Calculate population of molecular state for vibrations at 100, 600 cm⁻¹ and 3000cm⁻¹ at T=300K.

Answer:

$$\Phi_i = B \Omega A_i' = B (A_2/d^2) A_i'$$

$$2.0 \mu\text{W cm}^{-2} = B [0.010\text{cm}^2 / (50\text{cm})^2] \times 0.010\text{cm}^2$$

$$\text{Source radiance: } B = 5 \times 10^7 \mu\text{W cm}^{-2}\text{sr}^{-1}$$

Plus!
a

Slightly different variation of question a, note energies are 100, 600, 3000 cm⁻¹

$$n_i = [n_i g_i e^{-E_i/kT}] / Z(T)$$

$$n_i / n_0 = e^{(-E_i - E_0)/kT} = e^{(-h \nu / kT)} = e^{(-hc / \lambda kT)}$$

population of molecular state for vibrations at 100 cm⁻¹:

$$\begin{aligned} n_i / n_0 &= e^{(-hc / \lambda kT)} = e[(-6.626 \times 10^{-34} \text{Js} \times 3.00 \times 10^8 \text{ms}^{-1} \times 100 \text{cm}^{-1}) / 1.38 \times 10^{-23} \text{JK}^{-1} \times 300 \text{K}] \\ &= e[(-6.626 \times 10^{-34} \text{Js} \times 3.00 \times 10^8 \text{ms}^{-1} \times 10000 \text{m}^{-1}) / 1.38 \times 10^{-23} \text{JK}^{-1} \times 300 \text{K}] \\ &= 0.6188 \end{aligned}$$

population of molecular state for vibrations at 600 cm⁻¹:

$$\begin{aligned} n_i / n_0 &= e^{(-hc / \lambda kT)} = e[(-6.626 \times 10^{-34} \text{Js} \times 3.00 \times 10^8 \text{ms}^{-1} \times 600 \text{cm}^{-1}) / 1.38 \times 10^{-23} \text{JK}^{-1} \times 300 \text{K}] \\ &= 0.0561 \end{aligned}$$

population of molecular state for vibrations at 3000 cm⁻¹:

$$\begin{aligned} n_i / n_0 &= e^{(-hc / \lambda kT)} = e[(-6.626 \times 10^{-34} \text{Js} \times 3.00 \times 10^8 \text{ms}^{-1} \times 3000 \text{cm}^{-1}) / 1.38 \times 10^{-23} \text{JK}^{-1} \times 300 \text{K}] \\ &= 5.574 \times 10^{-7} \end{aligned}$$

2-11

$$3V * 0.25A = 0.75W \quad \text{power consumed} \quad \text{output} = 7.5 \times 10^{-3} W \text{ light}$$

a. Photons/sec: $550 \text{ nm} = 18,200 \text{ cm}^{-1}$

photons need conversion to energy units compatible with W

$$E = h\nu = hc\tilde{\nu} = (6.63 \times 10^{-34} \text{ J}\cdot\text{s}) (3.0 \times 10^{10} \text{ cm/s}) (1.82 \times 10^4 \text{ cm}^{-1})$$

$$= 3.61 \times 10^{-19} \text{ J}$$

photon rate: $(N/t) = 7.5 \times 10^{-3} W / 3.61 \times 10^{-19} \text{ J}$

$$= 2.08 \times 10^{16} \text{ phot/sec}$$

b. Cross section $10 \text{ cm}^2 \rightarrow \text{flux} = \phi_p = 2.08 \times 10^{15} \text{ phot/cm}^2 \cdot \text{sec}$

Vol: $1 \text{ m}^3 = (100)^3 \text{ cm}^3 = 10^6 \text{ cm}^3$

length of cylinder $10 \text{ cm}^2 = A \Rightarrow 10^5 \text{ cm} = l$

time to fill cylinder w/ photons: $t = 10^5 \text{ cm} / 3 \times 10^{10} \text{ cm/s}$

$$c = l/t \quad t = l/c = 0.33 \times 10^{-5} \text{ s}$$

total photons in volume: $\phi_p \cdot t \cdot A = 2.08 \times 10^{15} \frac{\text{ph}}{\text{cm}^2 \cdot \text{sec}} \cdot 0.33 \times 10^{-5} \text{ s} \cdot 10 \text{ cm}^2$

$$= 6.86 \times 10^9 \text{ photons/cm}^2 \cdot 10 \text{ cm}^2$$

$$= 6.86 \times 10^{10} \text{ photons in vol.}$$

John did much nicer: $\frac{n}{V} = \frac{\phi \cdot dt}{A \cdot dl} = \frac{\phi}{Ac} = \frac{2.08 \times 10^{16}}{10^{-3} \text{ m}^2 \cdot 3 \times 10^8 \text{ m/s}}$

$$= 6.86 \times 10^{10} \text{ ph/m}^3$$

c. Assume losses at $10 \text{ cm}^2 = A$

complex: $\phi_p = 2.08 \times 10^{15} \text{ ph/cm}^2 \cdot \text{sec}$

simple power: $P = 7.5 \times 10^{-3} \text{ W/A}$

$$\phi = \frac{7.5 \times 10^{-3} \text{ W}}{10 \text{ cm}^2} = \frac{7.5 \times 10^{-4} \text{ W/cm}^2}{3.61 \times 10^{-19} \text{ J}}$$

$$\phi_p = (hc\tilde{\nu}) = 2.08 \times 10^{15} \frac{\text{ph}}{\text{cm}^2 \cdot \text{s}} = 7.52 \times 10^{15} \text{ W/cm}^2$$

2-8

Thermal equilibrium $n_1 = n_0 e^{-\Delta E/kT}$

$T = 3000^\circ\text{C} = 3273\text{K}$ $kT = 1.38 \times 10^{-23} \text{ J/K} \cdot 3273\text{K} = 4.5 \times 10^{-20}$

$\Delta E = h\nu = h \frac{c}{\lambda} = 6.63 \times 10^{-34} \frac{\text{J}}{\text{s}} \cdot 3 \times 10^8 \text{ m/s} / 400 \times 10^{-9} \text{ m}$

~~Population excited:~~ $= 4.97 \times 10^{-19} \text{ J}$

$(n_1/n_0) = \exp(-4.97 \times 10^{-19} / 4.5 \times 10^{-20}) = 1.69 \times 10^{-5}$

2-12

$\Phi = 1.0 \times 10^{-3} \text{ W}$ $d = 2 \times 10^{-3} \text{ m} = 0.2 \text{ cm}^2$

$A = \pi r^2 = \pi d^2/4 = 3.14 \cdot 4 \times 10^{-6} \text{ m}^2/4 = 3.14 \times 10^{-6} \text{ m}^2$
 $= 3.14 \times 10^{-2} \text{ cm}^2$

Energy density is question but I thought I asked for irradiance

$E = \frac{\Delta\phi}{\Delta A} \approx \frac{\Phi}{A} = \frac{1.0 \times 10^{-3} \text{ W}}{3.14 \times 10^{-2} \text{ cm}^2} = 3.2 \times 10^{-2} \frac{\text{W}}{\text{cm}^2}$

Energy density is $U = \frac{\Delta Q}{\Delta V} \approx Q/V$

in a volume \rightarrow cylinder of diam 2mm

in 1s \rightarrow 1mJ \rightarrow $h = 3 \times 10^{10} \text{ cm long}$

$U = \frac{1.0 \times 10^{-3} \text{ J}}{9.4 \times 10^8 \text{ cm}^3} = 1.06 \times 10^{-12} \text{ J/cm}^3$ $V = 3 \cdot 3.14 \times 10^8 \text{ cm}^3 = 9.4 \times 10^8 \text{ cm}^3$

old available some method

PLUS Q

CO₂ population - not written well - except p.1. does this for 3 vibs as ratios to ground state $n_1/n_0, n_2/n_0$

ask for 3 temps - just change T in formula

3 vibs - just change $h\nu$ in formula

but population is: $n = n_{000} + n_{100} + n_{010} + n_{001} + n_{200}$ etc easier if use $k \sim 2/3 (\text{cm}^{-1}/\text{K})$ and assume negligible error

Consider $\nu = 1380 \text{ cm}^{-1}$: $n_{010}/n_{000} = \exp[-1680 / (0.67 \times 1000)] = 0.081$
 $= \exp[-1680 / (0.67 \times 300)] = 2.3 \times 10^{-4}$
 $= \exp[-1680 / (0.67 \times 77)] = 7.2 \times 10^{-15}$

etc.

Plus
b

Again not well-written problem

Loop: $1\text{ mm} \times 5\text{ mm}$ filament $\Rightarrow A = 5\text{ mm}^2$

50 W Color Temp: 2000 K

old problem
same method
Wien: $\lambda_{\text{max}} = \frac{c_2}{4.96 T} = \frac{2.897 \times 10^4}{T} = 1.448 \times 10^3 \text{ nm}$

or 1448 nm

(near IR)

or 1.45 μm

Stefan-Boltzmann: $M^b = \pi B^b = \pi \int_0^\infty B_\lambda^b d\lambda = \sigma T^4 = 5.67 \times 10^{-12} \text{ W cm}^{-2} \text{ K}^{-4} \times T^4$

if 90% is light: Total radiant emittance of bb:

$$5.67 \times 10^{-12} \cdot (2 \times 10^3)^4 = 90.7 \text{ W cm}^{-2}$$

Area: $5 \times 10^{-2} \text{ cm}^2$ — problem — emit all directions

$$\text{so } A \sim 3.14 \times 5 \times 10^{-2} \sim 16.7 \times 10^{-2}$$

radiant flux — $90.7 \text{ W cm}^{-2} \times 16.7 \times 10^{-2} \text{ cm}^2 \sim 15 \text{ W}$

(then correct 90% \sim 13.5 W)

new

b — color temp 1500 K

$$\lambda_{\text{max}} = \frac{c_2}{4.96 T} = \frac{2.897 \times 10^4}{T} = 1931 \text{ nm}$$

$$\text{total power: } M = \sigma T^4 = 5.67 \times 10^{-12} \text{ W cm}^{-2} \text{ K}^{-4} = 28.7 \text{ W cm}^{-2}$$

area: $3 \times 5 \text{ mm}^2 = 0.15 \text{ cm}^2$ but get 70%

$$\underline{I = 3 \text{ W}} \quad \text{— radance (intensity)}$$

Part II INCOHERENT LIGHT SOURCES

Chapter 4

4-1

#1 Usually when one buys a radiation source, the manufacturer will supply spectral radiance data so that one can judge if the intensity is sufficient for a given application. For tungsten lamps, the spectral radiance can be estimated from blackbody parameters.

- (a) Calculate the spectral radiance of a tungsten lamp at 500 nm with a color temperature of 2700 K, $\epsilon_\lambda = 0.40$, and $T_w = 0.92$ in $\text{W sr}^{-1} \text{cm}^{-2} \text{nm}^{-1}$ and photons $\text{s}^{-1} \text{cm}^{-2} \text{sr}^{-1} \text{nm}^{-1}$.
- (b) Apply propagation of error mathematics to the blackbody equation to determine how the uncertainty in temperature relates to an uncertainty in radiance. Evaluate the standard deviation and RSD in radiance at 500 nm caused by a 5°C temperature standard deviation under the conditions in part (a).

Answer:

(a) Radiance of tungsten lamp:

$$\begin{aligned} B_\lambda &= \epsilon(\lambda) T_w(\lambda) B_\lambda^b \\ &= \epsilon(\lambda) T_w(\lambda) c_1 \lambda^{-5} / (e^{c_2/\lambda T} - 1) \\ &= 0.40 \times 0.92 \times 1.190 \times 10^{16} \text{ W nm}^4 \text{cm}^{-2} \text{sr}^{-1} \times (500 \text{ nm})^{-5} / (e^{1.4388 \text{E}7 \text{ nm K} / (500 \text{ nm} \times 2700 \text{ K})} - 1) \\ &= 4.3792 \times 10^{15} \times (500)^{-5} / 42271.8 \text{ W sr}^{-1} \text{cm}^{-2} \text{nm}^{-1} \\ &= 3.315 \times 10^{-3} \text{ W sr}^{-1} \text{cm}^{-2} \text{nm}^{-1} \end{aligned}$$

The energy of a photon: $E = hc/\lambda = (6.63 \times 10^{-34} \text{ Js} \times 3.00 \times 10^8 \text{ ms}^{-1}) / (500 \times 10^{-9} \text{ m}) = 3.978 \times 10^{-19} \text{ J}$

$$\begin{aligned} B_\lambda &= 3.315 \times 10^{-3} \text{ W sr}^{-1} \text{cm}^{-2} \text{nm}^{-1} / 3.978 \times 10^{-19} \text{ J photon}^{-1} \\ &= 3.315 \times 10^{-3} \text{ Js}^{-1} \text{sr}^{-1} \text{cm}^{-2} \text{nm}^{-1} / 3.978 \times 10^{-19} \text{ J photon}^{-1} \\ &= 8.33 \times 10^{15} \text{ photons s}^{-1} \text{cm}^{-2} \text{sr}^{-1} \text{nm}^{-1} \end{aligned}$$

(b) $B_\lambda(2700\text{K}) = 3.315 \times 10^{-3} \text{ W sr}^{-1} \text{cm}^{-2} \text{nm}^{-1}$

$$\begin{aligned} B_\lambda(2695\text{K}) &= \epsilon(\lambda) T_w(\lambda) B_\lambda^b \\ &= \epsilon(\lambda) T_w(\lambda) c_1 \lambda^{-5} / (e^{c_2/\lambda T} - 1) \\ &= 0.40 \times 0.92 \times 1.190 \times 10^{16} \text{ W nm}^4 \text{cm}^{-2} \text{sr}^{-1} \times (500 \text{ nm})^{-5} / (e^{1.4388 \text{E}7 \text{ nm K} / (500 \text{ nm} \times 2695 \text{ K})} - 1) \\ &= 4.3792 \times 10^{15} \times (500)^{-5} / 43112.87 \text{ W sr}^{-1} \text{cm}^{-2} \text{nm}^{-1} \\ &= 3.250 \times 10^{-3} \text{ W sr}^{-1} \text{cm}^{-2} \text{nm}^{-1} \end{aligned}$$

$$\begin{aligned} B_\lambda(2705\text{K}) &= \epsilon(\lambda) T_w(\lambda) B_\lambda^b \\ &= \epsilon(\lambda) T_w(\lambda) c_1 \lambda^{-5} / (e^{c_2/\lambda T} - 1) \\ &= 0.40 \times 0.92 \times 1.190 \times 10^{16} \text{ W nm}^4 \text{cm}^{-2} \text{sr}^{-1} \times (500 \text{ nm})^{-5} / (e^{1.4388 \text{E}7 \text{ nm K} / (500 \text{ nm} \times 2705 \text{ K})} - 1) \\ &= 4.3792 \times 10^{15} \times (500)^{-5} / 41445.55 \text{ W sr}^{-1} \text{cm}^{-2} \text{nm}^{-1} \\ &= 3.381 \times 10^{-3} \text{ W sr}^{-1} \text{cm}^{-2} \text{nm}^{-1} \end{aligned}$$

$$\begin{aligned} \text{Average } B_\lambda(2695\text{K}, 2700\text{K}, 2705\text{K}) &= [B_\lambda(2695\text{K}) + B_\lambda(2700\text{K}) + B_\lambda(2705\text{K})] / 3 \\ &= [3.250 \times 10^{-3} + 3.315 \times 10^{-3} + 3.381 \times 10^{-3}] / 3 \\ &= 3.315 \times 10^{-3} \text{ W sr}^{-1} \text{cm}^{-2} \text{nm}^{-1} \end{aligned}$$

$$\text{Standard Deviation: } \sigma = \sqrt{\sum (B_\lambda - B_{\lambda \text{average}})^2 / (n-1)} = 6.550 \times 10^{-5}$$

$$\text{RSD} = \sigma / B_{\lambda \text{average}} = 6.550 \times 10^{-5} / 3.315 \times 10^{-3} \times 100\% = 1.976\%$$

H-2

Ruby 10 cm long
 $1.58 \times 10^{19} \text{ cm}^{-3}$ conc
 $1.27 \times 10^{-20} \text{ cm}^2$ crosssection 694 nm

a. (p35) defines $A = 0.434 \sigma b n$ or $\alpha = \sigma b n$
 $T = e^{-\alpha} = \exp[-(1.27 \times 10^{-20} \text{ cm}^2)(10 \text{ cm})(1.58 \times 10^{19} \text{ cm}^{-3})]$
 $= e^{-(2.0066)} = \underline{\underline{0.134}}$

b. Threshold population inversion and number density upper level need

$$(n_j - n_i)_{th} = \frac{\ln(1/\rho_1 \rho_2)}{2\sigma b} \quad \text{eqn 4-18}$$

derived from gain

$\rho_1 = 0.99, \rho_2 = 0.74, b = 10 \text{ cm}$
but σ is problem - can assume $\sigma =$ ground state value
but that would be a two level system
so not very realistic

$$(n_j - n_i)_{th} = \frac{\ln(1/(0.99 \cdot 0.74))}{2(1.27 \times 10^{-20} \text{ cm}^2)(10 \text{ cm})}$$
$$= \underline{\underline{1.23 \times 10^{18} \text{ cm}^{-3}}}$$

This implies

$$n_j - n_i = \Delta n$$
$$n_j + n_i = n_T$$

$$2n_j = \Delta n + n_T$$
$$n_j = (1.23 \times 10^{18} + 1.58 \times 10^{19})/2 \text{ cm}^{-3}$$
$$= \underline{\underline{0.85 \times 10^{19} \text{ cm}^{-3}}}$$

Note not physically reasonable
problem is a bit defective

Part III LASER LIGHT SOURCES

Chapter 4

4-14

#14 Calculate the longitudinal mode separation for a 0.25 m laser in Hz and nm for lasing wavelength of 488 nm.

Answer:

Longitudinal mode are separated by frequency difference $\Delta \nu_n = c/2d = 3.00 \times 10^8 \text{ ms}^{-1} / (2 \times 0.25 \text{ m}) = 6.00 \times 10^8 \text{ Hz}$

1.4

1.4 If two levels at 300K are in thermal equilibrium with $n_2/n_1 = 1/e$, calculate the frequency of the transition from 1 → 2. In what part of the spectrum does this occur?

Answer:

$n_2/n_1 = e^{-(h\nu/kT)} = 1/e$

$e^{-(h\nu/kT)} = e^{-1} \rightarrow -h \Delta \nu / kT = -1$

$\Delta \nu = kT/h = (1.38 \times 10^{-23} \text{ Jk}^{-1} \times 300 \text{ K}) / 6.63 \times 10^{-34} \text{ Js} = 6.24 \times 10^{12} \text{ Hz}$

$\lambda = c/\Delta \nu = 3.00 \times 10^8 \text{ mS}^{-1} / 6.24 \times 10^{12} \text{ Hz} = 48.08 \text{ } \mu\text{m}$ (middle IR) $\sim 200 \text{ cm}^{-1}$ (Far-IR)

modify values method OK

2.0

2.0 Calculate the number of longitudinal modes that occur in $\Delta \lambda = 1 \text{ nm}$ at $\lambda^0 = 1.06 \text{ } \mu\text{m}$ for a 1 m long laser cavity.

Answer:

$\lambda^1 = \lambda^0 - 1 = 1.06 \text{ } \mu\text{m} - 1 \times 10^{-3} \text{ } \mu\text{m} = 1.059 \text{ } \mu\text{m} = 1059 \text{ nm}$

$\nu^0 = c/\lambda^0 = 3.00 \times 10^8 \text{ mS}^{-1} / 1.06 \text{ } \mu\text{m} = 3.00 \times 10^8 \text{ mS}^{-1} / 1.06 \times 10^{-6} \text{ m} = 2.8301887 \times 10^{14} \text{ Hz}$

$\nu^1 = c/\lambda^1 = 3.00 \times 10^8 \text{ mS}^{-1} / 1.059 \text{ } \mu\text{m} = 3.00 \times 10^8 \text{ mS}^{-1} / 1.059 \times 10^{-6} \text{ m} = 2.8328612 \times 10^{14} \text{ Hz}$

$\nu_n = \nu^1 - \nu^0 = 2.8328612 \times 10^{14} \text{ Hz} - 2.8301887 \times 10^{14} \text{ Hz} = 2.6725 \times 10^{11} \text{ Hz}$

$\nu_n = nc/2d \rightarrow n = 2\nu_n d/c = (2 \times 2.6725 \times 10^{11} \text{ Hz} \times 1 \text{ m}) / (3.00 \times 10^8 \text{ mS}^{-1}) = 1.78 \times 10^3$

modify values method OK

4.4

Question 4.4: Ar+ Ion laser (from Kansas State site)

The difference between adjacent modes in Ar+ Ion laser is 100MHz. The mirrors are at the end of the laser tube.

Calculate: 1. The length of the laser cavity.

2. The mode number of the wavelength 488[nm].

3. The change in difference between adjacent modes when the tube is shortened to half its length.

Answer:

1. $\Delta \nu_n = c/2d \rightarrow d = c/2 \Delta \nu_n = 1.50 \text{ m}$

2. $\nu_n = c/\lambda = 3.00 \times 10^8 \text{ ms}^{-1} / 488 \text{ nm} = 6.15 \times 10^{14} \text{ Hz}$

$m = 2\nu_n d/c = (2 \times 6.15 \times 10^{14} \text{ Hz} \times 1.50 \text{ m}) / 3.00 \times 10^8 \text{ ms}^{-1} = 6.15 \times 10^6$

3. $m = 2\nu_n d'/c = (2 \times 6.15 \times 10^{14} \text{ Hz} \times 0.75 \text{ m}) / 3.00 \times 10^8 \text{ ms}^{-1} = 3.075 \times 10^6$

Thanks to Ling Wu for answers